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Roseline Esi Amoah Ghana Standards Authority, rosarchie@yahoo.com

Faustina Dufie Wireko-Manu Kwame Nkrumah University of Science and Technology

Ibok Oduro Kwame Nkrumah University of Science and Technology

See next page for additional authors

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Authors

Roseline Esi Amoah, Faustina Dufie Wireko-Manu, Ibok Oduro, Firibu Kwesi Saalia, William Otoo Ellis, Alexander Dodoo, Charles Dermont, and Maame Ekua Manful

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Research Article

Effects of Pretreatment and Drying on the Volatile Compounds of Sliced Solar-Dried Ginger (*Zingiber officinale* Roscoe) Rhizome

Roseline Esi Amoah (),¹ Faustina Dufie Wireko-Manu (),² Ibok Oduro (),² Firibu Kwesi Saalia (),³ William Otoo Ellis (),² Alexander Dodoo (),¹ Charles Dermont (),⁴ and Maame Ekua Manful ()⁵

¹Ghana Standards Authority, Accra PMB MB 245, Ghana

²Department of Food Science and Technology, Kwame Nkrumah University of Science and Technology, Kumasi, Ghana ³Department of Food Preserve Engineering, P. O. Par LC 77, University of Change Learne, Acare, Change

³Department of Food Process Engineering, P. O. Box LG 77, University of Ghana, Legon, Accra, Ghana

⁴Graduate School of Agriculture and Bioengineering, JUNIA ISA, Lille, France

⁵School of Food Science and Environmental Health, Technological University Dublin, Dublin, Ireland

Correspondence should be addressed to Roseline Esi Amoah; rosarchie@yahoo.com

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Ginger (*Zingiber officinale* Roscoe) rhizomes are mostly used as spice and medicine due to their high aroma intensity and medicinal bioactive compounds. However, the volatile compounds of ginger, partly responsible for its aroma and medicinal properties, can be affected by the pretreatment, drying method, and extraction processes employed. The objective of this study was to assess the effects of pretreatment and drying on the volatile compounds of yellow ginger variety at nine months of maturation. The effect of potassium metabisulfite (KMBS) and blanching pretreatment and drying on the volatile compounds of ginger using head space solid-phase microextraction with GCMS/MS identification (HS-SPME/GCMS/MS) was investigated. KMBS of concentrations 0.0 (control), 0.1, 0.15, 0.2, and 1.0% and blanching at 50°C and 100°C were used for pretreatment and dried in a tent-like concrete solar (CSD) dryer and open-sun drying (OSD). The different concentrations of KMBS-treated fresh ginger rhizomes did not result in any particular pattern for volatile compound composition identification. However, the top five compounds were mostly sesquiterpenes. The 0.15% KMBS-treated CSD emerged as the best pretreatment for retaining α -zingiberene, β -cubebene, α -farnesene, and geranial. The presence of β -cedrene, β -carene, and dihydro- α -curcumene makes this study unique. The 0.15% KMBS pretreatment and CSD drying can be adopted as an affordable alternative to preserve ginger.

1. Introduction

Ginger (*Zingiber officinale* Roscoe) is a rhizome of the aromatic Zingiberaceae family mainly used as a spice [1]. It possesses strong pungent [2] and unique flavour components which are dominated by α -zingiberene, α -curcumene, α -farnesene, β -sesquiphellandrene, β -bisabolene, geranial, and neral or Citral [3]. It is also used as herbal medicine due to its phenolic components such as gingerol, paradol, shogaol, and zingerone [4]. The amounts of these flavour components vary widely depending on the variety and processing methods used on the rhizome, especially the extraction and detection methods [5]. The subterranean environment of the rhizome before harvest

subjects it to microbial contamination [6], and this requires pretreatment in addition to regular cleaning protocols to reduce the microbial load [1]. Several pretreatment methods involving the use of both chemical (sodium hydroxide (NaOH)), sodium chloride (NaCL), potassium hydroxide (KOH), sulfites, sodium metabisulfites (NaMBS), and potassium metabisulfites (KMBS)) and physical (thermal blanching (hot water, steam, superheated steam, etc.)) treatment have been applied to foods. In addition, nonthermal treatments such as ultrasound and freezing [1] have been applied to agricultural produce during processing such as drying with varying effects on the product characteristics and on consumer acceptability of the final product. Apart from reducing microbial load, pretreatment of agricultural produce is frequently done to minimize colour and or texture changes during drying [1]. Blanching of ginger has been done at a temperature of 70°C for a maximum duration of 30 min which reduced the pungent compound, gingerol, drastically [7]. The ginger volatile oils containing the flavour components are also reduced upon heat treatment just as the gingerol responsible for the pungency [8, 9]. A pretreatment method that has frequently been used for agricultural produce including ginger without heat application is the use of dilute potassium metabisulfite (KMBS). Washing in dilute potassium metabisulfite has the additional advantage of reducing the microbial load as well as increasing the antioxidant activity of the product. It also partly preserves the ginger flavour compounds [10].

In addition to producing dried ginger of acceptable microbial safety, colour, and texture, the pretreatment and drying methods used must also preserve the volatile compounds. Flavour components in any food product are affected by the drying method [9, 11]. Harsh drying methods that use elevated temperatures beyond 60°C may negatively affect the flavour of ginger [8, 9, 12]. Hence, the use of pretreatment and drying methods in the processing of ginger rhizomes must be carefully selected to achieve dried ginger which contains the dominant volatile compounds.

The methods for extraction and analyses of volatile compounds are also important. Solid-phase microextraction (SPME) is a simple, sensitive, and solvent-free technique which integrates sampling, isolation, and concentration of analytes from samples into a single-step process making it a fast and inexpensive sampling method [13]. Head space solid-phase microextraction (HS-SPME) is a method used in determining the flavour compounds of several products including ginger oil. Yang et al. [14] analysed oven-dried ginger by HS-SPME and detected chiefly α -zingiberene (26.4%), sesquiphellandrene (10.2%), β -phellandrene (10.0%), camphene (7.6%), geranial (6.6%), α -curcumene (6.0%), and β -bisabolene (5.4%). Similar work repeated with varying oven temperatures and drying methods, such as microwave drying, freeze drying, silica gel drying, and vacuum drying, detected similar top 5 volatile compounds but with varying concentrations [3, 11].

In most African countries and other tropical developing economies, industrial drying mostly involves the use of electrical energy which comes with additional cost, while most traditional approaches use open-sun drying. However, the solar drying approach could be an alternative process relative to the opensun drying. Solar drying has the advantage of reducing microbial load and environmental contamination associated with opensun drying [15]. Nonetheless, there is a paucity of data on the effects of solar drying and pretreatment on the flavour profile of ginger rhizomes. Thus, this study assessed the effects of pretreatment (soaking in dilute solutions of potassium metabisulfite (KMBS) and blanching) and drying on the volatile compounds of yellow ginger variety at nine months of maturation.

2. Materials and Methods

2.1. Source of Raw Materials. Eighty kilograms (80 kg) of fresh yellow ginger rhizomes at nine months of maturity were purchased from an outgrower in the ginger

producing areas of Nkawie in the Ashanti region of Ghana. This is a town situated on latitude of 6° 39′ 59.99″ N and longitude of -1° 48′ 59.99″ W. The fresh ginger rhizomes were transported (6 hours) to the laboratory packed in a perforated nylon sack (to facilitate aeration).

2.2. Washing and Pretreating Fresh Ginger. The fresh raw ginger was soaked in water for two hours to remove the adhering sand/debris and washed vigorously three times each time with fresh water. The washed ginger rhizomes were sliced manually with a kitchen knife to a thickness of 3-5 mm. The sliced ginger was washed again in water. The sliced ginger was divided into seven parts; four parts were soaked separately in 0.1%, 0.15%, 0.2%, and 1.0% potassium metabisulfite (KMBS) concentrations for 5 minutes. Another one part was soaked in water for 5 minutes serving as control. The final two parts were blanched at 50°C and 100°C temperatures for 5 minutes and 60 seconds, respectively. The pretreated sliced ginger rhizomes were drained separately and sampled for analyses of the volatile compounds' composition using HS-SPME/GCMS.

2.3. Drying of Fresh Washed Sliced Ginger. The remaining sliced ginger of each pretreatment was divided into two parts: one part was dried in the concrete solar dryer (CSD) (Figure 1) and the other part was dried in the open sun (OSD) for five days. The temperatures and humidity ranged 29.0-52.50°C; 27-91.5% RH, and 28.0-55.5°C; 29.5-90.5% RH for the CSD and OSD, respectively, using a digital temperature-humidity data longer (HOBO pro v2 digital logger (Model U23-001, USA) (Appendix 1). The dried sliced ginger was further analysed for volatile compounds composition using HS-SPME/GCMS.

2.4. Preparation of Ginger for the Head Space Solid-Phase Microextraction (HS-SPME) Analysis

2.4.1. Pulverising and Milling of Ginger. All pretreated and untreated fresh ginger rhizomes were pulverised using a Moulinex blender (DPA141/35H-0517 R, France) for approximately 45 s and used for analysis. Pretreated and untreated sliced dried ginger were milled to pass through a sieve size of 1 mm using Moulinex mill (DPA141/35H-0517 R, France). The pulverized fresh and milled dried ginger samples were stored in plastic vials at -18° C until used for analysis.

2.4.2. HS-SPME/Gas Chromatography-Mass Spectrometry (GC-MS) Analysis. The choice of an SPME fibre is essential because it may affect the type of compounds isolated. According to Ding et al. [11], a 100 μ m polydimethylsiloxane (PDMS) fibre proved to be the most useful in the extraction of ginger volatiles when compared to 30 μ m PDMS, 75 μ m carboxen polydimethylsiloxane (Carboxen PDMS), and



FIGURE 1: Pictorial representation of the concrete solar dryer (CSD).

TABLE 1: Major volatile compounds detected in pretreated sliced fresh yellow ginger rhizome.

		Fraction (%)							
Retention time (min)	Compound name	Potassium metabisulfite concentration					Blanching temperature		
		Control	0.10%	0.15%	1%	0.20%	50°C	100°C	
8.69	α-Pinene					1.346			
8.98	Camphene	0.008	2.644	2.139	1.786	5.178	0.029		
10.85	α -(E)-Sabinene hydrate	0.015	4.136	3.417	3.706	6.196	0.050	0.099	
15.16	Neral					1.351			
15.18	cis-Z-α-bisabolene epoxide			1.235					
15.19	2,6-Octadienal, 3,7-dimethyl (Z)						0.032		
15.73	Geranial	0.008	2.647	2.231		3.156	0.037	0.052	
15.74	Dill-apiole			2.298					
21.99	α-Curcumene	0.014	4.192	32.920	4.294	16.827	0.488	0.665	
22.18	Germacrene D		2.301				0.030	0.051	
22.22	α-Cubebene	0.015			2.600	3.519			
22.61	α-Zingiberene		32.983		13.935				
22.62	Dihydro-α-curcumene	99.768			33.943	5.718		97.898	
22.87	α-Farnesene	0.040	12.149	11.103			0.154	0.266	
23.03	α-Bisabolene	0.040	5.885		8.089	10.686	98.821		
23.07	α-Bisabolene			7.648					
23.20	(+)-Epibicyclosesquiphellandrene				0.983				
23.48	δ-Carene			15.441					
23.52	α-Cedrene	0.046	15.614	4.171	16.213	21.543	0.201	0.159	
24.75	γ-Gurjunene	0.006	0.945		0.723				
26.17	Zingerone							0.037	

65 μm polydimethylsiloxane divinylbenzene (PDMS-DVD). The volatile components of ginger samples were extracted by head space solid-phase microextraction (HS-SPME) with a 100 μm length PDMS fibre column (Supelco, USA). Preliminary analysis on this fibre gave an optimum temperature of 50°C at 23 min extraction time for α-zingiberene, α-farnesene, α- curcumene, β-bisabolene, and β-cedrene.

Approximately 1 g of pulverised fresh ginger or milled dried ginger was weighed into a 10 mL amber glass vial. The fibre of $100 \,\mu\text{m}$ diameter was manually inserted into the septum of the amber glass vial in the hot air oven at 50°C for 23 min and allowed for adsorption. The fibre was removed after adsorption and manually injected into the Gas Chromatography-Mass Spectrometry (GCMS) (Varian CP-3800/3380, Walnut Creek, CA 94598 United States of America).

The Gas Chromatography coupled to Mass Spectrometry (GC-MS/MS) was used to identify the major compounds of the dried ginger. A Varian VF-1 ms capillary column $(30 \text{ m} \times 0.25 \text{ mm})$ attached to a Varian CP 3800 gas chromatography and a Varian Saturn 2000 spectrometry was used for the separation of the various components. The temperature gradient was as follows: 40°C (3 min), 130°C at 7°C/min (3 min), 220°C at 3°C/min (2 min), and 260°C at 10°C/min (5 min). The flow rate of the helium gas was 1 mL/ min and the fibre was introduced and injected at 230°C without split (split off). The compounds were identified by comparing the mass spectra of the chromatographs with built-in libraries with authentic standards. The relative percentage amount of each component was calculated by comparing its average peak area with the total area of the dry weight base.

		Fraction (%)								
Retention time (min)	Compound name	Potassium metabisulfite concentration					Blanching temperature			
		0.0%	0.10%	0.15%	0.20%	1%	50°C	100°C		
8.69	α-Pinene						4.860	0.003		
8.98	Camphene	2.9685	1.32	0.005	0.007					
10.82	α-Phellandrene	4.3375								
10.85	α -(E)-Sabinene hydrate		2.96	0.011			3.034	0.006		
15.16	Neral		1.70			0.002		0.002		
15.18	Z,Z,Z-4,6,9-Nonadecatriene			0.005	0.007					
15.19	2,6-Octadienal, 3,7-dimethyl (Z)						0.001			
15.55	Borneol					0.001				
15.73	Geranial	2.5272	2.61	11.718	0.017	3.122	0.003	0.003		
15.74	Dill-apiole			0.005						
21.29	α-Farnesene	0.0055								
21.99	α-Curcumene	5.9425	4.89		0.008	4.888	0.006	6.433		
22.18	Germacrene D	2.5849	2.86	0.081	0.115		2.964			
22.22	α-Cubebene			31.813	44.703	3.462		0.003		
22.61	α-Zingiberene			19.377	0.028					
22.62	Dihydro-a-curcumene	39.719	41.40			41.017	43.035	44.523		
22.87	α-Farnesene	15.536	16.20			17.073	16.459	17.137		
23.03	α-bisabolene	9.2911	9.69			10.194	10.130	10.963		
23.07	(6R,7 R)-bisabolene			36.948	55.046					
23.20	Epibicyclosesquiphellandrene					0.002				
23.52	α-Cedrene	17.073	19.14			20.231	19.490	20.911		
24.75	γ-Gurjunene				0.012					
26.17	Zingerone				0.004					

TABLE 2: Major volatile compounds detected in pretreated sliced solar dried yellow ginger rhizome.

3. Results and Discussion

The major volatile compounds identified are presented in Tables 1–3 and the HS-SPME/GCMS chromatograms in Appendix 2. The fresh control retained chiefly dihydro- α -curcumene (99.78%), which is quite different compared to the compounds retained in the KMBS-pretreated fresh ginger rhizomes. Potassium metabisulfite-treated fresh ginger samples recorded α -Curcumene, α -Zingiberene, β -Cedrene, σ -Carene, α -Bisabolene, and β -Bisabolene in different quantities as shown in Table 1.

The preservation of dihydro- α -curcumene (99.78%) as the major component in the fresh sample was totally different from the findings of similar work done [11] that detected compounds such as α -zingiberene (28.12%), β -citral (15.71%), α -farnesene (6.90%), β - sesquiphellandrene (7.65%), and α -curcumene (2.84%). The difference may be attributed to the differences in methodology and the source of the ginger. On the other hand, the samples treated with KMBS shared similar volatile compounds, especially 0.1% which had higher α -Zingiberene (33%) and α -Farnesene (12.15%) compared to the findings of Ding et al. [11]. Again, the retaining of high dihydro- α -curcumene which has the same molecular weight and formula as α -zingiberene in the control (0.0% KMBS) may be due to harsh extraction conditions and possibly the temperature. α -zingiberene is a sesquiterpene compound, unique to ginger, and was retained in 0.1% and 1.0% KMBS-treated fresh ginger in quantities of 33% and 13.94%, respectively. The use of different concentrations of KMBS to pretreat the ginger

rhizome did not show any particular pattern in the isolation of volatile compounds as the concentrations increased from 0.1% to 1.0%; however, the presence of α -zingiberene in higher quantities in the 0.1% KMBS-treated fresh ginger makes it more superior than the other concentrations and even the control of this study [16]. This shows that KMBS pretreatment had more diversified top five volatile compounds than the control.

The control of the fresh sample behaved similarly to the blanched samples by detecting one major volatile compound. The 100°C sample retained dihydro-α-curcumene as the major component (97.9%) just as the control, while the 50°C blanched sample had β -Bisabolene as the major component (98.8%). This could affect the flavour of the blanched samples immensely because the compounds preserved in a ginger sample together contribute to its unique flavour. The blanching temperature of 50°C and 100°C were chosen based on the loss of gingerol beyond heat application of 60°C [12]; also the traditional processors boil the ginger rhizomes for 10 min to cure it before drying, hence 100°C to ascertain its effect. The preservation of one major dominant volatile compound in the fresh and blanched ginger rhizome of this study is similar to the findings of some work done [17] retaining zerumbone (85-87%) as the single major volatile compound in fresh ginger rhizomes using the hydrodistillation method.

The effect of drying on the volatile compounds is shown in the differences between the fresh control and the control solar dried and open-sun-dried samples. The fresh control detected only dihydro- α -curcumene (99.8%) as shown in

		Fraction (%)							
Retention time (min)	Compound name	Potassium metabisulfite concentration					Blanching temperature		
		Control	0.10%	0.15%	0.20%	1%	50°C	$100^{\circ}C$	
8.69	α-Pinene								
8.98	Camphene	2.1371	0.003	4.292	5.547	0.003	1.333	0.003	
10.89	trans-4-Thujanol	4.1767	5.338	5.893	8.300	0.006	1.992	4.866	
15.16	Neral	0.0014							
15.19	2,6-Octadienal, 3,7 -dimethyl (Z)		0.002				0.001	0.001	
15.55	Borneol								
15.73	Geranial	2.6124	0.003	0.003	0.004	0.003	0.001	0.002	
21.99	α-Curcumene	5.9455	6.917	6.052	9.391	0.009	2.656	5.681	
22.18	Germacrene D				0.004			0.003	
22.22	α-Cubebene	2.7886	0.003	0.003		0.004	1.149		
22.61	1,3, 6,10-Dodeca tetraene,3, 7,11-trimethyl (Z, E)-			40.171					
22.61	α-Zingiberene	39.641							
22.62	Dihydro-α-curcumene		41.941		6.326	58.085	17.546	42.915	
22.87	α-Farnesene	15.57	16.653	15.126	25.222	0.023	62.812	16.487	
23.03	α-Bisabolene	9.3904	10.091	9.633	15.520	14.397	4.250	10.215	
23.52	α-Cedrene	17.72	19.032	18.813	29.651	27.440	8.252	19.811	
36.52	Caryophyllene oxide				0.004	0.002			

TABLE 3: Major volatile compounds detected in pretreated sliced open-sun dried yellow ginger rhizome.

Table 1, but the solar dried and open-sun-dried samples showed differences in the top 5 compounds detected. In the CSD samples, compounds detected were dihydro- α -curcumene (39.72%), α -farnesene (15.54%), β -bisabolene (9.69%), α -curcumene (5.94%), and β -phellandrene (4.34%) as shown in Table 2.

The detected compounds were different from the findings of one study that [14] reported mainly zingiberene, β -sesquiphellandrene, α -farnesene, and β -bisabolene or α -curcumene from air-dried ginger rhizome using the same fibre but different temperature and extraction time. The CSD samples also showed remarkable differences between the findings of some studies done [3, 11]. They reported that drying increased or decreased certain volatile compounds affecting the overall flavour. Irrespective of the drying method employed in their study, similar dominant volatile compounds were isolated but in different quantities. Therefore, the observed differences in findings of this study relative to the other studies may be attributed to differences in extraction time and temperature [14] and/or variety of ginger. The results also showed differences between the CSD and OSD compounds, with about 40% of α -zingiberene (39.64%) being retained in OSD. The presence of this compound is the mark of quality making the difference quite remarkable. The following compounds were also isolated: α -curcumene (5.95%), α -farnesene (15.57), β -bisabolene (9.39%), and trans-4-thujanol (4.18%) (Table 3). Thus, the characteristics of the OSD samples in terms of active compound composition were relatively more like the findings of Yang et al. [14], in retaining compounds such as α -zingiberene (53.12%), curcumene (4.9%), α -farnesene (8.61), β -bisabolene (5.98%), and β -sesquiphellandrene (13.03%), than when compared to the CSD samples. The differences were in the quantities which may be due to the variety and age of ginger rhizome or the differences in

method of extraction as well as the drying methods used [5, 9, 18]. Apart from the top 5 compounds, CSD and OSD shared Camphene (2.97%; 2.145%) and Geranial (2.53%; 2.61%) in similar quantities.

The combination of pretreatment and drying affects the flavour compounds. KMBS is used in preserving the colour [19] of a product and to reduce and/or prevent microbial proliferation. However, the residual sulfur dioxide (SO₂) may affect the final flavour [20] underscoring the need for the application of the right concentration. In this study, the results showed that drying and pretreatment affected the volatile compounds isolated, which gives the unique final flavour of the dried ginger rhizome. The 0.1% and 1.0% KMBS-treated samples showed minor differences in the top 5 compounds isolated from the fresh and dried samples. These concentrations showed only one compound which was found in the fresh but not in the dried samples. α -zingiberene (33%) was isolated in the fresh 0.1% KMBS samples but not in the dried samples. This may have decomposed to dihydro- α -curcumene of 41.40% for CSD and of 41.94% for OSD, respectively (Bartley and Foley, 1994). Considering the detected volatile compounds for the 0.1% KMBS-treated samples, the dried samples had higher quantities than the fresh samples such as β -cedrene (15.6; 19.14; 19.03%), α-farnesene (12.15; 16.20; 16.65%), β-bisabolene (5.89; 9.96; 10.09%), and α-curcumene (4.19; 4.89; 6.92%) for the fresh, CSD, and OSD, respectively. This agrees with some findings [11] that drying causes the reduction or increase of certain volatile compounds. For the 1.0% KMBS samples, dihydro- α -curcumene, β -cedrene, β - bisabolene, and β -carene were common to the fresh and dried samples but in different quantities as shown in Tables 1-3. However, in this treatment, while α -Zingiberene (33.9%) was isolated in the fresh samples, α -farnesene was peculiar to the dried samples. Both α -zingiberene and α -farnesene are sesquiterpenes with the same molecular formula and weight. Therefore, the effect of both KMBS and drying may account for this difference.

The volatile compounds of the 0.15% and 0.2% KMBS-treated samples showed a different pattern relative to the 0.1% and 1.0% KMBS-treated samples. The 0.15% and 0.2% KMBS-treated samples had more differences in the detected compounds. In these treatments, the fresh samples had more similar compounds to the OSD samples than the CSD. α -Bisabolene was the only common compound present in the fresh (7.65%) and CSD (36.95%) for the 0.15% KMBS-treated but in the 0.2% KMBS-treated samples, and the fresh samples shared no common compound with the CSD as shown in Tables 1 to 3. The 0.15% fresh samples shared β -cedrene (4.17; 18.81%), α -farnesene (11.10; 15.13%), and α -curcumene (32.9; 6.05%) with the OSD, respectively. The 0.15% KMBS-treated CSD was the only KMBS dried sample that isolated α -Zingiberene (19.38%) apart from the OSD dried control which is a mark of good quality ginger oil because it gives the specific ginger aroma [21] as well as compounds such as α -bisabolene (36.95%) and geranial (11.72%) which give ginger rhizome the "citrus-like" flavor (Bartley and Foley, 1994). This study reports that 0.15% treated CSD sample is more superior to all dried samples irrespective of the drying method or pretreatment used. Even though the OSD control may be considered superior for having higher α -zingiberene (39.64%), research showed that this particular treatment had the highest yeast and mould contamination [1] which is unsafe for human consumption.

Blanching is a heat treatment that is used mostly to reduce drying time and microbial load. However, the high heat treatment regime for blanching can cause a change in the chemical composition of ginger rhizome, reducing both the pungency and flavour or altering it greatly [7]. Blanching and drying had a great effect on the volatile compounds. Blanching at 50°C for 5 min and 100°C for 60 s for the fresh samples detected only one major volatile compound which is α -Bisabolene (98.82%) and dihydro- α -curcumene (97.90%), respectively. On the other hand, the blanched, CSD, and OSD samples showed similar top 5 compounds, namely, dihydro- α -curcumene, β -cedrene α -farnesene, β -bisabolene, and α -curcumene in spite of the temperature differences. The CSD 50°C samples showed similar compounds in almost equal quantities relative to the CSD 100°C samples (Table 2). This implies that the temperature of the blanched samples for CSD did not make much difference in the volatile compound composition as well as the quantities for both temperatures (dihydro- α -curcumene (43.04; 44.52%), β -cedrene (19.49; 20.91%), α -farnesene (16.45; 17.14%), β -bisabolene (10.13; 10.96%), and α -curcumene (4.89; 6.43%) for 50°C and 100°C, resp.). The pattern was different for the OSD samples for different temperatures as they gave different quantities of volatile compound (Table 3).

On the whole, none of the blanched samples, fresh or dried, retained α -zingiberene, which is the mark of quality ginger oil using HS-SPME. The absence of α -zingiberene from the blanched samples in this study reduces the quality of the samples and as such it is not a recommended choice of pretreatment for ginger.

The presence of β -cedrene, β -carene, and dihydro- α -curcumene isolated among the top 5 volatile compounds for the yellow variety makes this study unique from most work done using HS-SPME [3, 11, 14].

4. Conclusion

The effect of pretreatment with KMBS resulted in the appearance and disappearance of certain volatile compounds compared to the control. The different concentrations of KMBS treated with fresh ginger did not result in any particular pattern. However, α -zingiberene was detected from 0.1% to 1.0% treated samples, which is the mark of quality ginger oil, making 0.1% superior to a higher α -zingiberene fraction and recommended for fresh ginger preservation. Both blanched samples of 50°C and 100°C detected only dihydro- α -curcumene and β -bisabolene, which may affect the overall flavour of the product. Drying alone showed OSD and CSD samples as having similar volatile compounds, yet OSD was seen to be better for detecting a higher α -zingiberene fraction. On the other hand, the combination of pretreatment with KMBS and drying or blanching and drying showed the appearance and disappearance as well as increase and/or decrease of certain volatile compounds, which will determine the flavour of the dried ginger rhizomes. The 0.15% KMBS-treated CSD emerged as a better dried pretreatment for retaining α -zingiberene, β -cubebene, α -farnesene, and geranial. None of the blanched dried samples detected *a*-zingiberene using HS-SPME/GCMS/MS and as such it is not a good pretreatment method for ginger rhizomes when flavour preservation is the objective. The tent-like concrete solar dryer can be adopted for commercial ginger drying with 0.15% KMBS pretreatment to preserve the volatile compounds of ginger. The limit of this study lies in the solar drying methods not being as controlled as mechanical dryers would have been. Nonetheless, the study was targeted at farmers and processors as a more affordable ginger preservation method than the mechanical dryers. It is expected that farmers and processors would adopt the use of this cost-effective method. It is also important to consider studies on suitable packaging materials in the future.

Data Availability

Data are deposited in a repository.

Conflicts of Interest

The authors declare no conflicts of interest for the content published in this manuscript.

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Supplementary Materials

Appendix 1: environmental conditions of the solar drying. Appendix 2: chromatograms of the volatile compounds of HS-SPME/GC-MS/MS analysis of ginger rhizome. (*Supplementary Materials*)

References

- R. E. Amoah, F. D. Wireko-Manu, I. Oduro, F. K. Saalia, and W. O. Ellis, "Effect of pretreatment on physicochemical, microbiological, and aflatoxin quality of solar sliced dried ginger (*Zingiber officinale* Roscoe) rhizome," *Journal of Food Science and Nutrition*, vol. 8, pp. 1–9, 2020.
- [2] R. Offei-Okyne, J. Patterson, L. T. Walker, and M. Verghese, "Processing effects on phytochemical content and antioxidative potential of ginger," *Zingiber officinale*). *Journal of Food and Nutrition Sciences*, vol. 6, pp. 445–451, 2015.
- [3] H. Baokang, W. Guowei, C. Zhiyong, and Q. Luping, "Effect of oven drying, microwave drying, and silica gel drying methods on the volatile components of ginger (*Zingiber* officinale Roscoe) by HS-SPME-GC-MS," *Drying Technology: International Journal*, vol. 30, no. 3, pp. 248–255, 2012.
- [4] R. E. Amoah, S. Kalakandan, F. D. Wireko-Manu, I. Oduro, F. K. Saalia, and E. Owusu, "The effect of vinegar and drying (Solar and Open Sun) on the microbiological quality of ginger (*Zingiber officinale* Roscoe) rhizomes," *Journal of Food Science* and Nutrition, vol. 8, pp. 1–8, 2020.
- [5] H. Yeh, C. Chuang, H. Chen, C. Wan, T. Chen, and L. Lin, "Bioactive components analysis of two various ginger (*Zin-giber officinale* Roscoe) and antioxidant effect of ginger extracts," *LWT- Journal of Food Science and Technology*, vol. 55, pp. 329–334, 2014.
- [6] E. E. Abdulkadir, A. A. Tahiya, N. Saif, Al-Bahry, and S. B. Charles, "Fungi and aflatoxins associated with spices in the Sultanate of Oman," *Journal of Mycopathologia*, vol. 155, pp. 155–160, 2002.
- [7] H. Gan, E. Charters, R. Driscoll, and G. Srzednicki, "Effects of drying and blanching on the retention of bioactive compounds in ginger and turmeric," *Journal of Horticulturae*, vol. 3, p. 13, 2016.
- [8] F. Irda, S. E. Waningish, and Sukrasno, "Heating effect of ginger (*Zingiber officinale* Roscoe) in content of volatile oil and oleoresin," *International Journal of Research in Pharmacy* and Science, vol. 5, no. 2, pp. 132–136, 2014.
- [9] I. H. Sellami, W. A. Wannes, I. Bettaieb et al., "Qualitative and quantitative changes in the essential oil of *Laurusnobilis* L. leaves as affected by different drying methods," *Journal of Food Chemistry*, vol. 126, no. 2, pp. 691–697, 2011.
- [10] A. Sangwan, A. Kawatra, and S. Sehgal, "Nutritional composition of ginger powder prepared using various drying methods," *Journal of Food Science & Technology*, vol. 51, no. 9, pp. 2260–2262, 2012.
- [11] S. H. Ding, K. J. An, C. P. Zhao, Y. H. Guo, and Z. F. Wang, "Effect of dying methods on volatiles of Chinese ginger, (*Zingiber officinale* Roscoe)," *Journal of Food and Bioproduct Processing*, vol. 90, pp. 515–524, 2012.
- [12] D. A. Balladin, O. Headley, I. Chang-Yen, and D. R. Mcgaw, "High pressure liquid chromatographic analysis of the main pungent principles of solar dried West Indian ginger

(Zingiber officinale Roscoe)," *Renewable Energy*, vol. 13, no. 4, pp. 531–536, 1998.

- [13] M. L. Bhavya and R. Ravi, "Development and validation of headspace Solid-Phase microextraction coupled with gas chromatography (HSSPME-GC) method for the analysis of zingiberzerumbet L," *Natural Product Research*, vol. 35, 2019.
- [14] Z. Yang, W. Yang, Q. Peng et al., "Volatile phytochemical composition of the rhizome of ginger after extraction by headspace solid-phase microextraction, petroleum ether extraction and steam distillation extraction," *Bangladesh Journal* of *Pharmacology*, vol. 4, pp. 136–143, 2009.
- [15] A. W. Deshmukh, M. N. Varma, C. K. Yoo, and K. L. Wasewar, "Investigation of solar drying of ginger (*Zingiber officinale*): empirical modelling, drying characteristics and quality studies," *Chinese Journal of Engineering*, vol. 2014, p. 7, Article ID 305823, 2014.
- [16] A. Toure and Z. Xiaoming, "Gas chromatoghraphic analysis of volatile components of Guinean and Chinese ginger oils (*Zingiber officinale*) extracted by steam distillation," *Journal of Agronomy*, vol. 6, no. 2, pp. 350–355, 2007.
- [17] H. R. Juliani, A. R. Koroch, J. E. Simon, J. Asante-Dartey, and D. Acquaye, "Chemistry and quality of fresh ginger varieties (*Zingiber officinale*) from Ghana," *ISHS Acta Horticulturae*, vol. 756, 2007.
- [18] C. Gardelli, V. Papageorgiou, A. Mallouchos, K. Theodosis, and M. Komaitis, "Essential oil composition of *Pistacia lentiscus* L. and *Myrtus communis* L.: evaluation of antioxidant capacity of methalonic extracts," *Journal of Food Chemistry*, vol. 107, pp. 1120–1130, 2008.
- [19] M. Okzan and B. Cemeroglu, "Desulphiting dried apricots by exposure to hot air flow," *Journal of the Science of Food and Agriculture*, vol. 82, pp. 1823–1828, 2002.
- [20] O. I. Onyemaobi and G. O. Williams, "Cytogenetic effects of two food preservatives, sodium metabisulphite and sodium benzoate on the tips of Allium cepa Linn," *Ife Journal of Science*, vol. 14, no. 1, pp. 155–165, 2012.
- [21] Z. Kamaliroostaa, L. Kamaliroosta, and A. H. Elhamirad, "Isolation and identification of ginger essential oil," *Journal of Food Biosciences and Technology*, vol. 3, pp. 73–80, 2013.
- [22] G. S. Geeta and T. K. R. Reddy, "Aspergillus flavus Link and its occurrence in relation to other mycoflora on stored spices," *Journal of Stored Products Research*, vol. 26, no. 4, pp. 211–213, 1990.
- [23] P. C. Onyenekwe and S. Hashimoto, "The composition of the essential oil of dried Nigerian ginger (*Zingiber officinale* Roscoe)," *Journal of European Food Research and Technology*, vol. 209, no. 6, pp. 407–410, 1999.