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Liang Jiang

Technological University Dublin

David Kennedy

Technological University Dublin, david.kennedy@tudublin.ie

Stephen Jerrams

Technological University Dublin, stephen.jerrams@tudublin.ie

See next page for additional authors

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Authors

Liang Jiang, David Kennedy, Stephen Jerrams, and Tony Betts

Enhancement of dielectric properties with the addition of bromine and dopamine modified barium titanate particles to silicone rubber

Liang Jiang^{1, a}, David Kennedy^c, Stephen Jerrams^a, Anthony Betts^b

^a Centre for Elastomer Research, Focas Research Institute, Dublin Institute of Technology, Dublin 8, Ireland

^b Applied Electrochemistry Group, Focas Research Institute, Dublin Institute of Technology, Dublin 8, Ireland

^c Department of Mechanical Engineering, Dublin Institute of Technology, Dublin 1, Ireland

Abstract

Dual coated barium titanate (BT) particles were prepared using dopamine in conjunction with bromine in order to enhance the dielectric constant of silicone rubber (SR) composites containing evenly distributed BT particles. The results showed that both dopamine and bromine were deposited on the BT particles. The dielectric constant of the SR/BT composite was significantly increased from 3.6 to 4.7 at 1 kHz with the addition of BT modified with dopamine (DP-BT). Moreover, the dielectric constant further rose to 4.9 at 1 kHz when the DP-BT particle was grafted with bromine (Br-DP-BT). This facile process represents an efficient and convenient way of enhancing the dielectric properties of dielectric elastomer (DE) composites.

Keywords: Coating; Barium titanate; Composites; Dielectric elastomer

1. Introduction

Dielectric elastomers (DEs) can respond to electrical stimuli by changing their shape and belong to the family of smart materials [1-3]. Generally, DEs can induce several levels of strains (from 10 to 300%), but they require large electric fields (usually around 100 V/ μm) to achieve large actuated strains [4]. It has been reported that the operating electric field of DEs could be decreased for these materials by increasing the DEs' dielectric constants and also through lowering Young's modulus [5]. A soft silicone rubber (SR) based DE has the advantages of low Young's modulus combined with lower viscoelasticity and good thermal stability [6]. However, its deformability under an electric field is limited with a low inherent dielectric constant (≈ 2.9) [2, 6]. The most popular approach to increasing the dielectric constant is by incorporating nano/micron sized high-dielectric filler particles into the DE matrix. Examples of fillers used are titanium dioxide (TiO_2) [7-9], lead magnesium niobate-lead titanate (PMN-PT) [10, 11], BT [12, 13] and functionalized graphene sheet (FGS) [14]. Ferroelectric oxides which exhibit a spontaneous electric polarization can also be reoriented in an external field. In particular, these oxides possess a large dielectric constant. Perovskite BT is one example of such oxides and exists as a white powder. It has a high dielectric constant of above 1200 [15]. Nevertheless, inorganic BT has poor compatibility with SR and this disadvantage prevents its further application in soft actuators. To negate this limitation, researchers focused on surface modification of BT [13, 16]. Dopamine, which can interact strongly with metal oxides by forming hydrogen bonds [17], was considered an effective chemical with which to modify the surface of BT particles [16, 18]. Moreover, the aromatic group, the hydrogen-bonded

¹Corresponding author, email: liang.jiang@mydit.ie

hydroxyl-dopamine species and the bromide group are of high polarisability [19-21] which is propitious to the enhancement of the dielectric constant. This research had the aim of enhancing the dielectric constant of SR based DE composites by adding the DP-BT and a further increase in the dielectric constant was sought by grafting bromine onto the DP-BT surface.

2. Experimental

2.1 DP-BT and Br-DP-BT preparation

Fig. 1 shows reactions for obtaining the DP-BT and the Br-DP-BT. The coating research on DP-BT was reported previously [22]. In this work, 10 g of BT was dispersed in 50 ml of ethanol-water mixture (with a mass ratio of 1:1) and stirred for 30 min. Thus, OH functional groups were added to the particle surfaces. The mixture was then poured into 450 ml of deionized water with the addition of 1 g of dopamine hydrochloride. Thereafter, the suspension was stirred at 60 °C overnight prior to being subjected to ultrasonic shaking for 30 min. The DP-BT coated particles were recovered by vacuum filtration. Subsequently they were washed using deionized water and finally dried at 60 °C for one day in a vacuum to avoid oxidization of the dopamine.

For deposition of bromine on DP-BT particles, an approach of directly grafting the bromine on dopamine was employed [23]. The resulting DP-BT particles were initially ground up and dissolved in a mixture of 20 ml acetic acid and 60 ml chloroform. Subsequently 1.25 ml concentrated bromine liquid was added and reacted overnight with magnetic stirring. The solution was heated at 60 °C in order to remove the remnant bromine, acetic acid and chloroform. Finally, the powder was washed with deionized water and dried at 60 °C for one day.

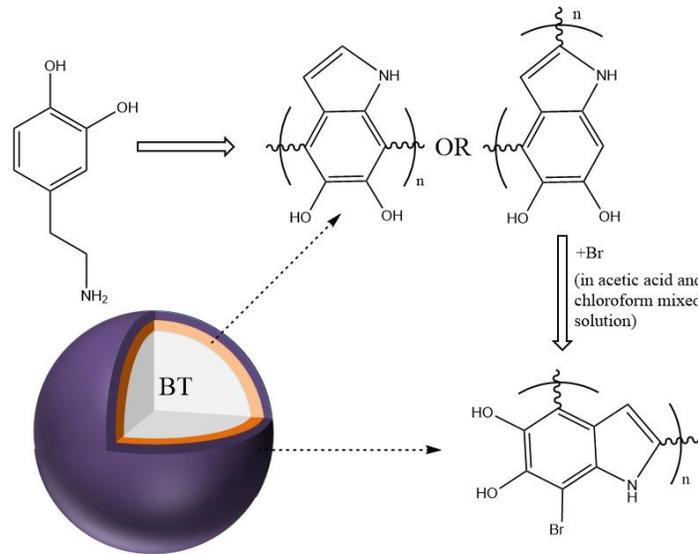


Fig. 1. Schematic diagrams of the fabrication of bromine and dopamine coated BT.

2.2 DE films preparation

Initially, the two part of SR were mixed at a weight ratio of 1:1 and dissolved in heptane. Then, the particles (BT, DP-BT and Br-DP-BT) with the required weight percentage of 10% were added to the mixture. In order to achieve uniform dispersion of the SR and fillers in heptane, the mixture

was subjected to ultrasonic shaking for 20 min. To complete the process, the mixture was poured into a watch glass, subjected to ultrasonic shaking for 30 min and heated for 3 h at 80 °C in a water bath with an ultrasonic cleaner which performed the function of heating and shaking under aerated conditions.

2.3 Characterisation

The morphological features of all particles in this work were observed using a Scanning Electron Microscope (SEM) (Zeiss Supra). Elemental analysis of these particles was carried out using an Energy Dispersive X-Ray Spectrometer (EDS) (Oxford Inca Xmax) which was coupled to the SEM. Fourier Transform Infrared Spectroscopy (FTIR) spectra of the micro-particles were recorded in the 350-4000 cm^{-1} range with a Perkin Elmer 400 Series Spectrometer. Dielectric measurements of DE films were conducted on a Turnkey broadband dielectric spectrometer at 20 °C in the frequency range of 10 Hz to 1 MHz.

3. Results and Discussion

Fig. 2 shows the SEM images of BT, DP-BT and Br-DP-BT particles. As shown in Fig. 2 (a) the BT particles were almost-spherical with a diameters of around 500 nm. They formed clusters easily and the surfaces of these particles were relatively smooth. After coating with dopamine and bromine, the treated particles increased in diameter by more than 200 nm (refer to Fig. 2 (b) and Fig. 2 (c)). Additionally, the surfaces of the coated particles became rough as a result of being modified with dopamine and bromine reagents.

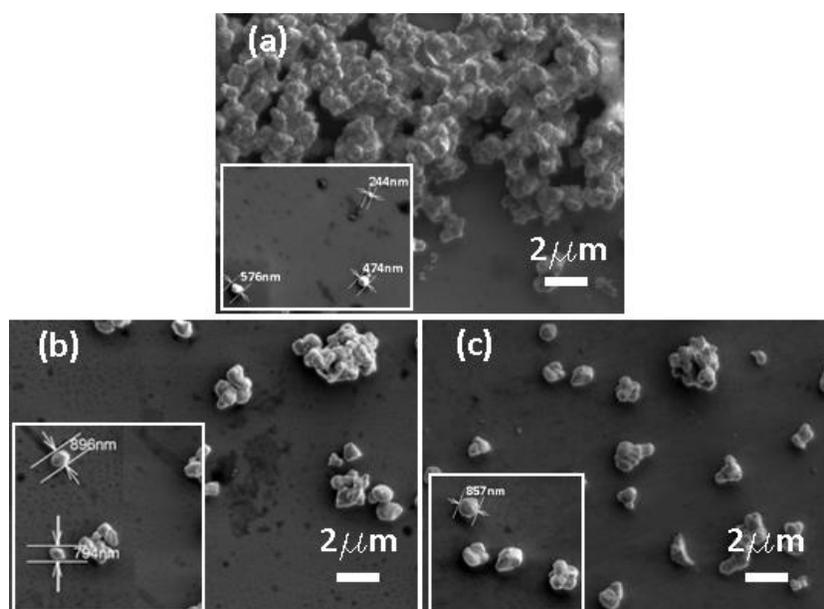


Fig. 2. SEM micrographs of particles used in the experiments: (a) BT; (b) DP-BT; (c) Br-DP-BT.

Fig. 3 shows the EDS spectrum of DP-BT and Br-DP-BT. Element Ba along with elements Ti and O can be observed in Fig. 3 (a) indicating the fundamental constituents of BT. Carbon (C) was also detected in each of the particles mainly due to residues present in various chamber surfaces such as vacuum pump and from sample surface migration and reaction with the electron beam

forming a carbon rich environment [24]. Moreover, the contents of C in DP-BT (46 wt%) and Br-DP-BT (44 wt%) are more than that in BT (35 wt%) indicating the deposition of dopamine on BT particles (refer to Figs. 3 (b) and (c)). However, the content of element O changed only slightly after modification with dopamine and bromine with the O atomic mass fraction of dopamine nearly equal to that of BT (21 wt%). Fig. 2 (c) shows the Br peak on the spectra indicating the deposition of Br on DP-BT particles.

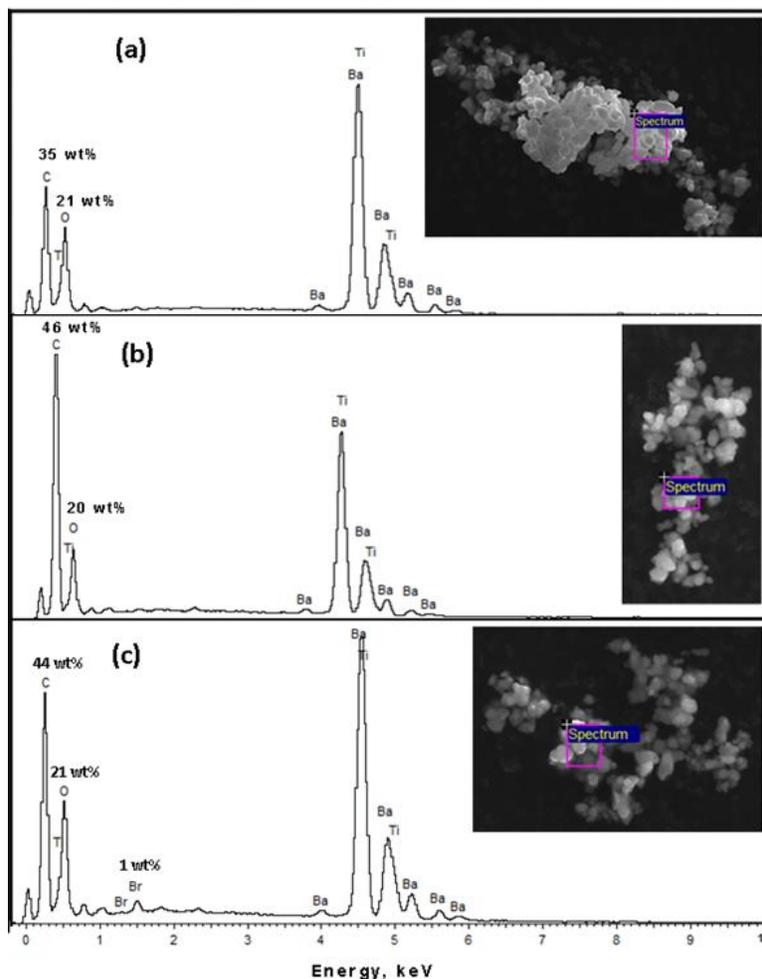


Fig. 3. EDS spectrum of DP-BT and Br-DP-BT.

Fig. 4 (a) shows the FTIR spectra of DP-BT and Br-DP-BT over a range of $500 - 600 \text{ cm}^{-1}$. The peaks at 512 and 524 corresponding to the C-Br stretch vibrations can be observed in the fingerprint region of the spectra. Fig.4 (b) shows the FTIR spectra of dopamine coated BaTiO_3 (DP-BT) and bromine grafted DP-BT (Br-DP-BT) in the functional group region ($1000 - 4000 \text{ cm}^{-1}$). Due to the deposition of dopamine on BaTiO_3 , the C-H bond appeared at 2916 cm^{-1} . The transmittance peaks located at 1607 cm^{-1} and 1286 cm^{-1} corresponded to aromatic amine N-H bending vibrations and C-N stretching vibrations respectively. In addition, the appearance of the peaks at 2568 cm^{-1} and 2168 cm^{-1} was interpreted as indicating the presence of the Ba-OH and Ti-

OH formation [25] which was probably due to the addition of ethanol during the fabrication of DP-BT.

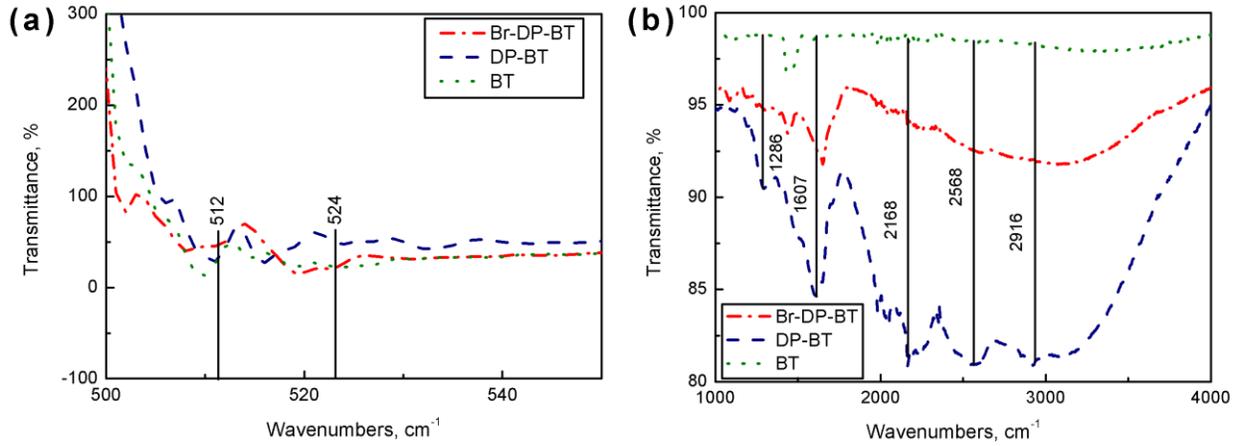


Fig.4. FTIR spectrum of DP-BT and Br-DP-BT in the frequency ranges of (a) 500-550 cm^{-1} ; (b) 1000 – 4000 cm^{-1} .

Fig. 5 shows the spectra of dielectric constant related to frequency for pure SR, SR/ 10wt% BT composite, SR/10 wt% DP-BT composite and SR/10 wt% Br-DP-BT composite. As can be observed from Fig. 4, pure SR displayed a dielectric constant of about 2.9 at 1 kHz. The dielectric constant increased over the entire frequency range with the addition of 10 wt% BT and to 3.4 at 1 kHz. After adding the DP-BT particles in the SR matrix, the dielectric constant rose at all testing frequencies, most significantly to 4.7 at 1 kHz. Moreover, the dielectric constant was further enhanced by about 0.2 when SR was filled with 10 wt% Br-DP-BT compared with that of SR/ 10 wt% DP-BT composite. However, the dielectric constant changed only slightly in the wide test frequency range for each SR composite, probably because of the amorphous structure presented in SR which prevents the dielectric constant being influenced from the dipole orientation in the electric field [26].

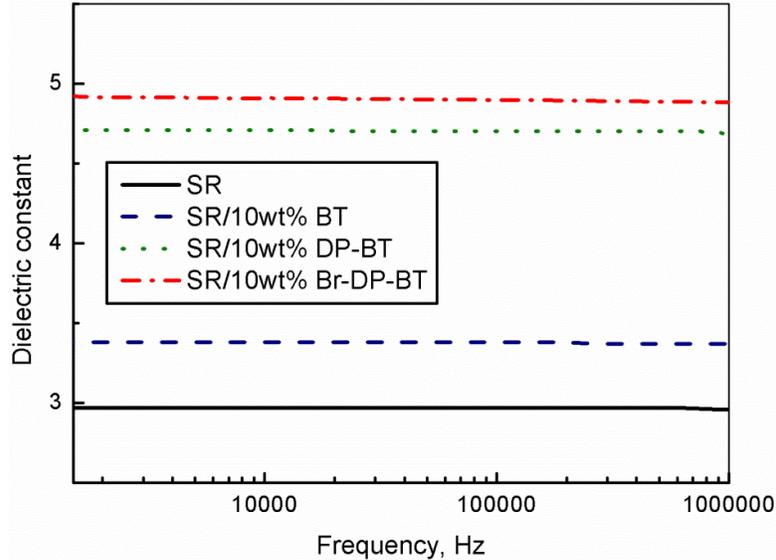


Fig. 5. Dependence of the dielectric constant on frequency.

Conclusions

This work illustrated an efficacious and relatively simple coating method for increasing the dielectric constants of SR/BT composites. SEM images showed the increases in diameter of the particles after they were coated with dopamine and bromine. It was clearly shown from the EDS and the FTIR spectrum results that dopamine and bromine were deposited on BT particles. Moreover, a dielectric constant of 4.9 was obtained by SR with 10 wt% Br-DP-BT which is numerically 2.5 higher than that of the SR/10 wt% BT composite. **Though, the dielectric constant of SR/10 wt% Br-DP-BT composite rose slightly (0.2) by comparison with the SR/10 wt% DP-BT, it was resumed that the dielectric constant should increase more significantly for higher filler loadings.**

Acknowledgements

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